Quality of Irrigation Water



Lecture note for Soil and Water Management Course Prepared by Dr ND Nang

Definition of water quality



• Water quality refers to the characteristics of a water supply that will influence its suitability for a specific use Quality is defined by certain physical, chemical and biological characteristics

Water Quality Problems

• Salinity

Salts in soil or water reduce water availability to the crop to such an extent that yield is affected.





Water Quality Problems

• Water infiltration rate

Relatively high sodium or low calcium content of soil or water reduces the rate at which irrigation water enters soil to such an extent that sufficient water cannot be infiltrated to supply the crop adequately from one irrigation to the next.



Water Quality Problems

Specific ion toxicity

Certain ions (sodium, chloride, or boron) from soil or water accumulate in a sensitive crop to concentrations high enough to cause crop damage and reduce yields.







Salinity

- Salts accumulate in the crop root zone to a concentration that causes a loss in yield
- Originate from:
 - A saline, high water table
 - Salts in the applied water

Salinity

• Mechanism of salt injury

The salt concentration in the root zone rise up to 2 to 5 times due to transpiration or by evaporation from the soil surface

- *Osmotic effect*: plant growth depression is attributable to a decrease in osmotic potential
- Specific ion effect: concentration of specific ions (Na, Cl, B) is greater than that expected from osmotic effect alone

Salinity

Guidelines for interpretations of water quality for irrigation

		Degree of Restriction on Use			
Potential Irrigation Problem	Units	None	Slight to Moderate	Severe	
Salinity (affects crowater availability)	р				
EC_w	dS/m	< 0.7	0.7 – 3.0	> 3.0	
(or)					
TDS	mg/L	< 450	450 - 2000	> 2000	

• Infiltration rate for the applied water or rainfall is appreciably reduced and water remains on the soil surface too long or infiltrates too slowly to supply the crop



- Mechanism
 - Slaking : Slaking is the breakdown of a lump of soil into smaller fragments on wetting. It is caused when clay swells and the trapped air bursts out
 - Dispersion : Dispersion (the separation of soil into single particles) is governed by soil texture, clay type, soil organic matter, soil salinity and exchangeable cations.

GUIDELINES FOR INTERPRETATIONS OF WATER QUALITY FOR IRRIGATION¹

			Units	Degree of Restriction on Use			
Potential Irrigation Problem		None		Slight to Moderate	Severe		
Infiltration (affects infiltration rate of water into the soil. Evaluate using EC_w							
and SAR together)							
SAR	= 0 - 3	and EC_{w}	=		> 0.7	0.7 - 0.2	< 0.2
	= 3-6		=		> 1.2	1.2 - 0.3	< 0.3
	= 6-12		=		> 1.9	1.9 – 0.5	< 0.5
	= 12 - 20		=		> 2.9	2.9 - 1.3	< 1.3
	= 20 - 40		=		> 5.0	5.0 - 2.9	< 2.9

$$SAR = \frac{(Na)}{\sqrt{\frac{(Ca + Mg)}{2}}}$$

- SAR : Sodium Adsorption Ratio
- (Na) and (Ca + Mg) are the concentrations of the soluble ions in meq/L

Ion toxicity

• Toxicity problems occur if certain constituents ions in the soil or water are taken up by the plant and accumulate to concentrations high enough to cause crop damage or reduced yields

Ion toxicity

GUIDELINES FOR INTERPRETATIONS OF WATER QUALITY FOR IRRIGATION

Degree of Restriction on Use

Potential Irrigation Problem	Units	None	Slight to Moderate	Severe
Specific Ion Toxicity (affects sensitive crops)				
Sodium (Na)				
surface irrigation	SAR	< 3	3 – 9	>9
sprinkler irrigation	me/L	< 3	>3	
Chloride (Cl)				
surface irrigation	me/L	< 4	4 - 10	> 10
sprinkler irrigation	me/L	< 3	>3	
Boron (B)	mg/L	< 0.7	0.7 - 3.0	> 3.0